

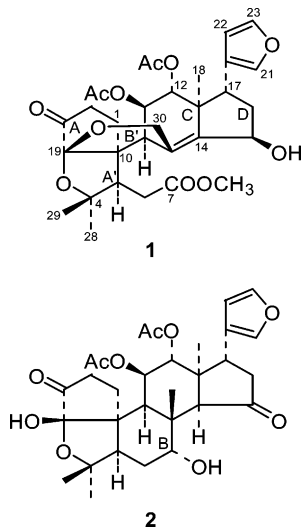
Musidunin and Musiduol, Insect Antifeedants from *Croton jatrophoides*Ken-ichi Nihei,[†] Yukihiro Asaka,[‡] Yoshihiro Mine,[§] Yoichi Yamada,[⊥] Masayuki Iigo,[†] Tadashi Yanagisawa,[†] and Isao Kubo*^{||}

Department of Applied Biochemistry, Faculty of Agriculture, Utsunomiya University, Utsunomiya, Tochigi 321-0943, Japan, Department of Life Sciences, Faculty of Science and Technology, Kinki University, Higashi-osaka, Osaka 577-8502, Japan, Life Science Research Institute, Kinki University School of Medicine, Sayama, Osaka 589-8511, Japan, Department of Chemistry, Faculty of Education, Utsunomiya University, Utsunomiya, Tochigi 321-0943, Japan, and Department of Environmental Science, Policy and Management, University of California, Berkeley, California 94720–31122

Received February 12, 2006

Two novel limonoids, musidunin (**1**) and musiduol (**2**), were isolated from a methanol extract of *Croton jatrophoides* by bioassay-guided fractionation. Their structures were established by extensive NMR experiments. Interestingly, A,B-seco limonoid **1** contains a unique acetal annulation of A, A', and B' rings. Both limonoids exhibited antifeedant activities against two pests, *Pectinophora gossypiella* and *Spodoptera frugiperda*.

Phytochemicals acting as deterrents to insect feeding are frequently found in pest-resistant plants, which can be lead compounds to develop environmentally friendly pest control agents.¹ In our continuing search for insect-affecting phytochemicals,^{2–4} a bitter plant, *Croton jatrophoides* Pax. (Euphorbiaceae), which has been used as a folk medicine in East Africa, was identified to possess a highly pest-resistant property. Seven limonoids as insect antifeedant principals were isolated from a methanol extract of *C. jatrophoides*, and their structures, having a remarkable A–A' ring closure, were determined by X-ray crystallographic and NMR analyses.^{5–8} Limonoids are attractive biologically active substances because they possess anti-HIV activity,⁹ antimalarial activity,¹⁰ and cytotoxicity against cancer cell lines^{11,12} in addition to insect antifeedant activity.^{13,14} Systematic investigation of phytochemicals in *C. jatrophoides* was performed using antifeeding assays against two pests, the larvae of *Pectinophora gossypiella* and *Spodoptera frugiperda*. As a result, musidunin (**1**) and musiduol (**2**) were isolated as active constituents. Structural determinations and antifeedant potencies of **1** and **2** are described in this paper.



The methanol extract of the root bark of *C. jatrophoides* was partitioned between H₂O and *n*-hexane, chloroform, ethyl acetate,

and *n*-butanol, in this order. The chloroform fraction showed strong insect antifeedant activity and was fractionated using silica gel column chromatography and preparative TLC. Compounds **1** and **2** were then isolated by preparative HPLC and were designated as musidunin and musiduol, respectively, based on the Swahili name of *C. jatrophoides*, “msinduzi”.

Musidunin (**1**) was isolated as an amorphous solid. The molecular formula was unequivocally established as C₃₁H₃₈O₁₁ from the molecular ion peak at *m/z* 586 [M⁺] by HREIMS analysis. The IR spectrum of **1** showed strong absorption near 1750 cm⁻¹, which indicated the presence of a carbonyl moiety. Aromatic methine signals (δ_{H} 7.34, 7.20, 6.23; δ_{C} 142.6, 140.0, 110.6) and a quaternary carbon signal at δ 122.6 in the ¹H and ¹³C NMR spectra were assigned to a β -furan ring. Hence, this compound is classified as a limonoid.

The COLOC experiment revealed that the β -furan ring was attached to a methine (δ_{H} 3.04; δ_{C} 43.4) in the D ring (Figure S1, Supporting Information). The methine signal was connected to an oxymethine proton at δ 4.80 (H-15) via diastereotopic methylene protons at δ 2.27 and 1.92 (H-16) in the ¹H–¹H COSY spectrum. These protons were correlated with a quaternary carbon at δ 49.1 (C-13) that was bonded to the α -oriented methyl at H-18 (δ_{H} 1.01; δ_{C} 17.1) and to a quaternary olefin carbon at δ 145.0 (C-14) in the COLOC spectrum. The hydroxyl group at C-15 was deduced as β -oriented due to the NOE correlation between H-15 and H-18 (Figure S2, Supporting Information). C-13 possessed a COLOC correlation with an acetylated oxymethine (H-12; δ_{H} 5.36; δ_{C} 75.3) in the C ring. This oxymethine was coupled with an acetylated oxymethine (H-11; δ_{H} 5.63; δ_{C} 69.1) that was further connected to a methane (H-9; δ_{H} 3.16; δ_{C} 39.3) in the ¹H–¹H COSY spectrum. H-9 and H-11 are α -oriented, as indicated by NOE correlations with H-18. The cross-peak between H-9 and an olefin quaternary carbon (δ_{C} 127.9) in the COLOC spectrum indicated that the tetrasubstituted olefin (C-8 and C-14) was in the C ring. The two acetylated oxymethines in the C ring are often found in limonoids from *C. jatrophoides*, and the vicinal coupling constants are usually 0.8–5.1 Hz.^{5–8} The large coupling constant (9.8 Hz) between H-11 and H-12 of **1** suggested that the vicinal acetoxy moieties possessed pseudoequatorial relationships on this cyclohexene ring.

The geminal dimethyls (δ_{H} 1.44, 1.30; δ_{C} 31.9, 26.1) identified by the ¹³C–¹H COSY experiment were placed at C-28 and C-29 in the A' ring as in dummin.⁷ In the COLOC experiment, H-28 was correlated to a methine (δ_{H} 2.81; δ_{C} 54.7) that was coupled with methylene protons at δ 2.59 and 2.31. This proton sequence was assigned to H-5 and H-6. However, from inspection of the COLOC correlations, C-6 was adjacent to a carboxymethyl moiety (δ_{H} 3.70; δ_{C} 52.1, 172.6) at C-7, indicating that the B ring in **1** was absent.

* To whom correspondence should be addressed. Tel: +1-510-643-6303. Fax: +1-510-643-0215. E-mail: ikubo@uclink.berkeley.edu.

[†] Faculty of Agriculture, Utsunomiya University.

[‡] Kinki University.

[§] Kinki University School of Medicine.

[⊥] Faculty of Education, Utsunomiya University.

^{||} University of California, Berkeley.

Diastereotopic methylene protons at δ 2.97 and 2.12 were connected to methylene protons at δ 2.57 in the ^1H – ^1H COSY spectrum. The signals of H-5, H-9, and its diastereotopic methylene (H-1) possessed cross-peaks to a quaternary carbon (C-10; δ_{C} 57.8). Also, COLOC correlations between the vicinal methylene signals (H-1 and H-2) and a carbonyl carbon (δ_{C} 210.7) and an oxygenated quaternary carbon (δ_{C} 108.8) confirmed the presence of the A ring. Thus, compound **1** contained the A and A' rings as in other dumsin-type limonoids,^{5,7} and the B ring was absent. The remaining NMR signals were due to an oxymethylene (δ_{H} 4.81, 4.31; δ_{C} 64.3). These methylene protons were correlated to olefin carbons (C-8 and C-14), the oxygenated quaternary carbon (C-19), and the methylene carbon (C-9). Thus, the oxymethylene was included in the B' ring. Annulation of the A', A, and B' rings of **1** was supported by several key NOE correlations.

Musidulol (**2**) was isolated as an amorphous solid, and the molecular formula was established as $\text{C}_{30}\text{H}_{38}\text{O}_{10}$ by HRFABMS. An absorption at 1757 cm^{-1} in the IR spectrum indicated the presence of a carbonyl moiety. The ^1H and ^{13}C NMR spectra of **2** were similar to those of duminin,⁷ except for absence of signals originating from a phenylacetoxymethoxy moiety in the C ring and the trisubstituted epoxide in the D ring. Instead, carbonyl carbon (δ_{C} 219.9), isolated methine (δ_{H} 2.85; δ_{C} 57.9), and oxymethine (δ_{H} 3.98; δ_{C} 71.3) signals were identified by ^{13}C NMR, DEPT, and ^{13}C – ^1H COSY experiments.

The oxymethine was assigned to H-7, as it was correlated to a methine (H-5; δ_{H} 2.61; δ_{C} 48.9) through a methylene (H-6; δ_{H} 2.22, 1.70; δ_{C} 28.2) in the ^1H – ^1H COSY spectrum. The NOE correlation between H-7 and H-30 and the small coupling constant of H-7 (3.3 Hz) indicated that the OH group was α -oriented (Figure S3, Supporting Information). The isolated methine possessed COLOC cross-peaks with C-8, H-30, and H-18, indicating that the methine was in the D ring (C-14). The remaining carbonyl carbon was assigned to C-15. The α -orientation of H-14 was deduced from the NOE between H-14 and H-9 and H-18.

Ring D, having a carbonyl moiety, has been found in other limonoids.^{16,17} As in **2**, and in other dumsin- and zumsin-type limonoids from *C. jatrophoides*, the D ring is resistant to oxidation. Moreover, C-30 and the B ring of **1** appear to be preferentially oxidized before A ring oxidation. Although many B ring oxidized limonoids have been reported,^{18–20} compound **1** is classified as one of a very few examples with respect to this type of acetal annulation of A, A', and B' rings. One possibility of biogenesis of **1** from dumsenin⁷ is illustrated in Figure S4 (Supporting Information).

Both limonoids **1** and **2** exhibited insect antifeedant activities ($\text{PC}_{50} = 3\text{ }\mu\text{g/mL}$, $\text{PC}_{95} = 10\text{ }\mu\text{g/mL}$; $\text{PC}_{50} = 4\text{ }\mu\text{g/mL}$, $\text{PC}_{95} = 20\text{ }\mu\text{g/mL}$, respectively) against the second-instar larvae of *P. gossypiella* in a leaf disk assay.²¹ Compound **2** also possessed insect antifeedant activity against *S. frugiperda* with $\text{PC}_{50} = 2.0\text{ }\mu\text{g/mL}$ and $\text{PC}_{95} = 36\text{ }\mu\text{g/mL}$. Insect antifeedant activity of **1** was also observed, although **1** could not be isolated in sufficient quantity for a detailed assay. The two pests *P. gossypiella* and *S. frugiperda* damage cottonseed, corn, and other commercial crops produced in countries such as the United States, India, China, and Australia.^{22,23} Therefore, unique limonoids such as **1** and **2** may be useful as pest control agents due to their potent insect antifeedant activities.

Experimental Section

General Experimental Procedures. Specific rotations were recorded in MeOH on a JASCO DIP-370 digital polarimeter (Tokyo, Japan). IR spectra were recorded in CHCl_3 on a Horiba FT-720 spectrometer (Kyoto, Japan). ^1H and ^{13}C NMR spectra were recorded in CDCl_3 with TMS as internal reference on JEOL JNM-GX-400 and EX-400 spectrometers (Akishima, Japan). HREIMS and HRFABMS were measured in the positive-ion mode on a JEOL JMS-700TKM spectrometer. Preparative TLC plates were purchased from Analtech, Inc. (Newark, DE). All solvents were purchased from Aldrich Chemical Co. (Milwaukee, WI). Preparative HPLC was performed with an

EYELA LPG-1000 instrument and an EYELA UV7000 detector (Tokyo Rikakikai Co. Ltd., Tokyo, Japan), on a $10\text{ mm} \times 250\text{ mm i.d.}$, $10\text{ }\mu\text{m}$, Alltech Econosil C_{18} column (Deerfield, IL). Initially, 40% MeCN/ H_2O was used as the HPLC solvent. The gradient elution was started at 5 min and proceeded from 40% to 80% MeCN/ H_2O in 30 min. For the isolation of **1**, isocratic elution with 45% MeCN/ H_2O was then performed. The flow rate and detected wavelength were adjusted to 5 mL/min and 210 nm, respectively.

Plant Material. Root bark of the East African medicinal plant locally known as "msinduzi" was collected near Mombasa, Kenya, and the plant was identified as *C. jatrophoides* (Euphorbiaceae).²⁴ A plant specimen (AC 76-134) was deposited in the Department of Botany herbarium at the University of Nairobi.

Extraction and Isolation. The root bark was removed at the collection site. The air-dried root bark (500 g) was cut into small pieces and extracted with MeOH (500 mL \times 3) at ambient temperature for 2 weeks. The solvent was evaporated in vacuo, and the resulting residue (40 g) was partitioned between H_2O (800 mL) and *n*-hexane (200 mL \times 3), CHCl_3 (200 mL \times 3), and EtOAc (200 mL \times 3), respectively. A leaf disk assay against second-instar larvae of *P. gossypiella* identified the CHCl_3 fraction (4.5 g) as containing the antifeedant activity. This fraction was further divided into six fractions (I: 0.2 g, II: 0.7 g, III: 0.8 g, IV: 0.4 g, V: 1.1 g, and VI: 0.5 g) using chromatography on silica gel (70–230 mesh, 250 g) eluted with 1–20% MeOH/ CHCl_3 . Subsequent bioassays showed moderate activity in the 20% MeOH/ CHCl_3 eluted fraction (V). This fraction was subjected to preparative TLC with 20% MeOH/ CHCl_3 and then further purification by preparative HPLC to give 1.8 mg of **1** ($t_{\text{R}} = 15.0$ min by the isocratic mode) and 2.5 mg of **2** ($t_{\text{R}} = 17.0$ min).

Musidunin (1): colorless solid; $[\alpha]_{\text{D}}^{24} -40.6$ (c 0.09, CHCl_3); IR (CHCl_3) ν_{max} 1747, 1371, 1245 cm^{-1} ; ^1H NMR (CDCl_3 , 400 MHz) δ 7.34 (1H, m, H-23), 7.20 (1H, bs, H-21), 6.23 (1H, m, H-22), 5.63 (1H, dd, $J = 9.8, 7.3$ Hz, H-11), 5.36 (1H, d, $J = 9.8$ Hz, H-12), 4.81 (1H, d, $J = 15.0$ Hz, H-30a), 4.80 (1H, d, $J = 7.8$ Hz, H-15), 4.31 (1H, d, $J = 15.0$ Hz, H-30b), 3.70 (3H, s, COOCH_3), 3.16 (1H, d, $J = 7.3$ Hz, H-9), 3.04 (1H, dd, $J = 12.4, 6.8$ Hz, H-17), 2.97 (1H, ddd, $J = 13.7, 9.3, 4.4$ Hz, H-1a), 2.81 (1H, dd, $J = 11.7, 3.4$ Hz, H-5), 2.59 (1H, m, H-6a), 2.57 (2H, m, H-2), 2.31 (1H, dd, $J = 16.1, 3.4$ Hz, H-6b), 2.27 (1H, ddd, $J = 14.6, 12.4, 7.8$ Hz, H-16a), 2.12 (1H, dt, $J = 13.7, 9.8$ Hz, H-1b), 2.05 (3H, s, OCOCH_3), 1.92 (1H, dd, $J = 14.6, 6.8$ Hz, H-16b), 1.75 (3H, s, OCOCH_3), 1.61 (1H, bs, OH), 1.44 (3H, s, H-28), 1.30 (3H, s, H-29), 1.01 (3H, s, H-18); ^{13}C NMR (CDCl_3 , 100 MHz) δ 210.7 (C, C-3), 172.6 (C, C-7), 171.43 (C, OCOCH_3), 171.39 (C, OCOCH_3), 145.0 (C, C-14), 142.6 (CH, C-23), 140.0 (CH, C-21), 127.9 (C, C-8), 122.6 (C, C-20), 110.6 (CH, C-22), 108.8 (C, C-19), 89.9 (C, C-4), 75.3 (CH, C-12), 69.1 (CH, C-11), 68.5 (CH, C-15), 64.3 (CH₂, C-30), 57.8 (C, C-10), 54.7 (CH, C-5), 52.1 (CH₃, COOCH_3), 49.1 (C, C-13), 43.4 (CH, C-17), 41.0 (CH₂, C-16), 39.3 (CH, C-9), 34.4 (CH₂, C-6), 34.3 (CH₂, C-2), 31.9 (CH₃, C-28), 30.1 (CH₂, C-1), 26.1 (CH₃, C-29), 21.0 (CH₃, OCOCH_3), 20.5 (CH₃, OCOCH_3), 17.1 (CH₃, C-18); HREIMS m/z 586.2394 $[\text{M}]^+$ (calcd for $\text{C}_{31}\text{H}_{38}\text{O}_{11}$, 586.2414).

Musidulol (2): colorless solid; $[\alpha]_{\text{D}}^{25} -24.2$ (c 0.12, CHCl_3); IR (CHCl_3) ν_{max} 1757, 1734, 1371, 1230 cm^{-1} ; ^1H NMR (CDCl_3 , 400 MHz) δ 7.38 (1H, m, H-23), 7.17 (1H, bs, H-21), 6.20 (1H, m, H-22), 5.40 (1H, d, $J = 2.8$ Hz, H-12), 4.91 (1H, t, $J = 2.8$ Hz, H-11), 3.98 (1H, d, $J = 3.3$ Hz, H-7), 3.67 (1H, t, $J = 9.2$ Hz, H-17), 3.56 (1H, bs, OH), 2.85 (1H, bs, H-14), 2.66 (1H, m, H-2a), 2.64 (1H, dd, $J = 11.2, 9.2$ Hz, H-16a), 2.61 (1H, dd, $J = 9.6, 5.4$ Hz, H-5), 2.50 (1H, d, $J = 2.8$ Hz, H-9), 2.48 (1H, dd, $J = 11.2, 9.2$ Hz, H-16b), 2.35 (1H, m, H-2b), 2.30 (1H, m, H-1a), 2.22 (1H, dd, $J = 13.0, 9.6$ Hz, H-6a), 2.06 (3H, s, OCOCH_3), 2.05 (3H, s, OCOCH_3), 1.75 (1H, ddd, $J = 14.0, 10.4, 2.0$ Hz, H-1b), 1.70 (1H, ddd, $J = 13.0, 5.4, 3.3$ Hz, H-6b), 1.61 (1H, bs, OH), 1.45 (3H, s, H-29), 1.40 (3H, s, H-30), 1.36 (3H, s, H-28), 0.78 (3H, s, H-18); ^{13}C NMR (CDCl_3 , 100 MHz) δ 219.9 (C, C-15), 210.0 (C, C-3), 171.0 (C, OCOCH_3), 169.5 (C, OCOCH_3), 143.6 (CH, C-23), 139.9 (CH, C-21), 122.9 (C, C-20), 110.4 (CH, C-22), 104.9 (C, C-19), 85.3 (C, C-4), 71.5 (CH, C-11), 71.3 (CH, C-7), 71.2 (CH, C-12), 57.9 (CH, C-14), 55.7 (C, C-10), 48.9 (CH, C-5), 44.5 (C, C-13), 43.5 (CH, C-9), 43.1 (CH₂, C-16), 39.5 (C, C-8), 37.9 (CH, C-17), 35.2 (CH₂, C-1), 32.0 (CH₃, C-29), 31.5 (CH₂, C-2), 28.2 (CH₂, C-6), 28.1 (CH₃, C-28), 21.9 (CH₃, OCOCH_3), 21.5 (CH₃, C-18), 20.9 (CH₃, OCOCH_3), 19.4 (CH₃, C-30); HRFABMS m/z 559.2545 $[\text{M} + \text{H}]^+$ (calcd for $\text{C}_{30}\text{H}_{39}\text{O}_{10}$, 559.2543).

Insect Antifeedant Assay. Second-instar larvae of *P. gossypiella* and *S. frugiperda* were used as test organisms, and a leaf disk assay was performed by the method reported previously.²¹ Briefly, leaf disks (1 cm²) were punched out from a glandless cotton cultivar, randomized, and arranged (12 disks/dish) concentrically on moistened filter paper within polyethylene foam grids inside glass Petri dishes (100 mm × 15 mm). Alternate disks were treated on their upper surface with either 25 μL of acetone or 0–100 μg of the sample dissolved in 25 μL of acetone applied with a microliter syringe. Three larvae were then placed in the dishes at 22 °C in a dark incubator. After 48 h, the larvae were removed and disks were examined visually. Percent area of the leaf disk consumed versus control was recorded. PC₅₀ and PC₉₅ values are the concentrations at which the test compounds afforded ca. 50% and 95% protection of the host plant substrate, respectively. The assays were performed in triplicate, on separate occasions, and their range of error was within 0.5 μg/mL.

Acknowledgment. We are indebted to the late Dr. J. A. Klocke for performing the leaf disk assay and to Mr. A. Chapya for collection and identification of the plant material. K.N. gratefully acknowledges the financial support for this project, in part, from JSPS Tropical Bio-resources Research Fund.

Supporting Information Available: Selected NOE and COLOC correlations for **1** and **2** and possible biogenetic pathway of **1** from dumsenin. This material is available free of charge via the Internet at <http://pubs.acs.org>.

References and Notes

- Isman, M. *Pestic. Outlook* **2002**, *13*, 152–157.
- Hammond, D. G.; Rangel, S.; Kubo, I. *J. Agric. Food Chem.* **2000**, *48*, 4410–4417.
- Nihei, K.; Shibata, K.; Kubo, I. *Phytochemistry* **2002**, *61*, 987–990.
- Kubo, I.; Kinst-Hori, I.; Nihei, K.; Soria, F.; Takasaki, M.; Calderün, J. S.; Céspedes, C. L. *Z. Naturforsch.* **2003**, *58C*, 719–725.
- Kubo, I.; Hanke, F. J.; Asaka, Y.; Matsumoto, T.; He, C. H.; Clardy, J. *Tetrahedron* **1990**, *46*, 1515–1522.
- Nihei, K.; Hanke, F. J.; Asaka, Y.; Matsumoto, T.; Kubo, I. *J. Agric. Food Chem.* **2002**, *50*, 5048–5052.
- Nihei, K.; Asaka, Y.; Mine, Y.; Ito, C.; Furukawa, H.; Ju-ichi, M.; Kubo, I. *J. Agric. Food Chem.* **2004**, *52*, 3325–3328.
- Nihei, K.; Asaka, Y.; Mine, Y.; Kubo, I. *J. Nat. Prod.* **2005**, *68*, 244–247.
- Sunthitikawinsakul, A.; Kongkathip, N.; Kongkathip, B.; Phonnakhu, S.; Daly, J. W.; Spande T. F.; Nimit, Y.; Napaswat, C.; Kasisit, J.; Yoosook, C. *Phytother. Res.* **2005**, *17*, 1101–1103.
- Omar, S.; Zhang, J.; MacKinnon, S.; Leaman, D.; Durst, T.; Philogene, B. J.; Arnason, J. T.; Sanchez-Vindas, P. E.; Poveda, L.; Tamez, P. A.; Pezzuto, J. M. *Curr. Top. Med. Chem.* **2003**, *3*, 133–139.
- Zhou, H.; Hamazaki, A.; Fontana, J. D.; Takahashi, H.; Wandscheer, C. B.; Fukuyama, Y. *Chem. Pharm. Bull.* **2005**, *53*, 1362–1365.
- Miller, E. G.; Porter, J. L.; Binnie, W. H.; Guo, I. Y.; Hasegawa, S. *J. Agric. Food Chem.* **2004**, *52*, 4908–4912.
- Nakatani, M.; Abdelgaleil, S. A.; Kassem, S. M.; Takezaki, K.; Okamura, H.; Iwagawa, T.; Doe, M. *J. Nat. Prod.* **2002**, *65*, 1219–1221.
- Carpinella, C.; Ferrayoli, C.; Valladares, G.; Defago, M.; Palacios, S. *Biosci. Biotechnol. Biochem.* **2002**, *66*, 1731–1736.
- Tchuendem, M. H. K.; Ayafor, J. F.; Connolly, J. D.; Sterner, O. *Tetrahedron Lett.* **1998**, *39*, 719–722.
- Mulholland, D. A.; Monkhe, T. V.; Coombes, P. H.; Rajab, M. S. *Phytochemistry* **1998**, *49*, 2585–2590.
- Cortez, D. A. G.; Fernandes, J. A.; Vieira, P. C.; da Silva, M. F. G. F.; Ferreira, A. G. *Phytochemistry* **2000**, *55*, 711–713.
- Luo, X. D.; Wu, S. H.; Wu, D. G.; Ma, Y. B.; Qi, S. H. *Tetrahedron* **2002**, *59*, 7797–7804.
- Saad, M. M. G.; Iwagawa, T.; Doe, M.; Nakatani, M. *Tetrahedron* **2003**, *59*, 8027–8033.
- Ismail, I. S.; Ito, H.; Hatano, T.; Taniguchi, S.; Yoshida, T. *Chem. Pharm. Bull.* **2004**, *52*, 1145–1147.
- Kubo, I. In *Methods in Plant Biochemistry*, Vol. 6; Hostettmann, K., Ed.; Academic Press: London, 1991; pp 179–193.
- Qain, M.; Zilberman, D. *Science* **2003**, *299*, 900–902.
- Marengo, R. J.; Foster, R. E.; Sanchez, C. A. *J. Econ. Entomol.* **1992**, *85*, 1285–1292.
- Kokwaro, J. O. *Medicinal Plants of East Africa*; East African Literature Bureau: Nairobi, Kenya, 1976; p 88.

NP060068D